TWO FLAVONE GLUCOSIDES FROM SIDERITIS LEUCANTHA

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In continuation of our work from *Sideritis leucantha* Cavanilles [1], two new flavone glucosides, 5,7,3',4'-tetrahydroxy-6,8-dimethoxyflavone 7-glucoside (1) and 5,7,4'-trihydroxy-6,8,3'-trimethoxyflavone 7-glucoside (2), have been identified.

UV data [2] indicated that there were free hydroxyls at the 5-, 3'- and 4'-positions in 1 and also that the 6- and/or 8hydroxyls were substituted [3]. Similarly, free 5- and 4'hydroxyls were confirmed in 2 as well as substituents in the 6- and/or 8-positions and two or more substituents in the B-ring of the flavone nucleus. Both substances gave glucose on acid hydrolysis. Permethylation of the aglycones of both 1 and 2 gave nobiletin (5,6,7,8,3',4'-hexamethoxyflavone) identified by co-TLC with an authentic marker. Alkaline degradation of 2 gave vanillic acid, identified by TLC with an authentic sample. The permethylated derivatives of both glucosides showed identical R_c on TLC. The MS spectra of PM and PDM derivatives suggest that peaks of m/e (A + H) – Me may correspond to the quinonoidal structures formed by the loss of Me from positions 6, 8 or 3'. Thus, the peak corresponding to the ion of m/e (A + H) – CD₃ seen on the PDM spectrum of 1 (which is not present in the PDM spectrum of 2) indicates a different type of substitution at the 3'-position of each derivative.

The determination of structures and the elucidation of the substitution pattern were carried out by standard spectroscopic and analytical methods [4-6]. From these data 1 was identified as 5,7,3',4'-tetrahydroxy-6,8-dimethoxyflavone 7-glucoside and 2 as 5,7,4'-trihydroxy-6,8,3'-trimethoxyflavone 7-glucoside. Chhabra et al. [7] have identified 5,7,3',4'-tetrahydroxy-6,8-dimethoxyflavone from the gums of Gardenia lucida and G. gummifera

and Horie et al. [8] have characterized 5,7,4'-trihydroxy-6,8,3'-trimethoxyflavone from unripe fruit of Citrus sudachi. However, the 7-glucosides of both these substances have not been reported previously.

EXPERIMENTAL

The MeOH extract [1] was successively chromatographed (Whatman No. 1) in the following solvent systems: 15% HOAc $(R_c s: 1 \text{ and } 2, 0.15), HOAc-HCO_2H-H_2O(10:2:3)(R_c s: 1 \text{ and } 2,$ 0.76), 50% HOAc (R_c s: 1 and 2, 0.74) and n-BuOH-EtOH-H₂O (4:1:2.2) (R_f 1: 0.72; R_f 2: 0.55). Mp 1 192–195°; mp 2 185–187°. R_f PM 1 = R_f PM 2, TLC (Sigel) EtOAc (R_f : 0.72), CHCl₃-EtOAc-Me₂CO (5:4:1) (R_f : 0.68). UV 1 λ_{max} nm: MeOH, 344, 278, 258; NaOMe, 406, 274; AlCl₃, 434, 366, 310 sh, 288; AlCl₃/HCl, 362, 310 sh, 282, 266: NaOAc, 410, 270; NaOAc/H₃BO₃, 366, 270. UV 2 λ_{max} nm: MeOH, 344, 272, 254; NaOMe, 408, 266; AlCl₃, 390 sh, 360, 300 sh, 278; AlCl₃/HCl, 390 sh, 360, 300 sh, 278; NaOAc, 410, 268. MS (70 eV direct inlet) m/e: PM 1 = PM 2, 606 (M⁺, 11), 388 (A + H = m, 100), 373 $(m - 15, 82), 255 (a_1 - 2, 9), 197 [(a_1 - 2) - CO, 31], 182 (14),$ $167(16), 165(b_1, 16), 162(b_2, 14), 218(T_1, 15), 187(T_2, 37), 155(T_3, 167)$ 39), 101 (175). MS (70 eV, direct inlet) m/e: PDM 1, 627 (M⁺, 7), 397 (A + H = m, 100), 382 (m - 15, 45), 379 (m - 18, 71), 288 $(a_1 - 2, 22), 200 [(a_1 - 2) - CO, 47], 185 (16), 171 (b_1, 22),$ 168 (b₂, 27), 230 (T₁, 27), 196 (T₂, 62), 161 (T₃, 52), 107 (142). MS (70 eV, direct inlet) m/e: PDM 2, 624 (M⁺, 7), 394 (A + H = m, 100, 379 (m - 15, 45), 368 (17), 339 (14), 313 (21), 285 (13),264 (13), 257 (12), 239 (23), 236 (19), 213 [(a₁ - 2) + 15, 17], 211(20), $200 \lceil (a_1 - 2) - CO, 23 \rceil$, 185 (31), 171 (33), $168 (b_1, 29)$, 165(b₂, 26), 196 (T₂, 43), 161 (T₃, 57), 107 (238).

Scheme 1. MS partial fragmentation of PDM of 1.

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FLAVONOIDS FROM ABRUS PRECATORIUS

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Abrus precatorius, having medicinal properties [1], has been examined for its chemical components [2–6]. This communication describes the isolation of a new flavone, abrectorin and a known glycoside, desmethoxycentaureidin 7-O-rutinoside.

A. precatorius seed kernels (2 kg) were defatted using petrol and then extracted exhaustively with EtOH. Solvent-free EtOH extract was repeatedly extracted with Et₂O and then EtOAc to separate non-glycosidic and glycosidic components. Combined Et₂O and EtOAc extracts containing non-glycosidic components were concentrated. The solvent-free concentrate was chromatographed on a Si gel column; elutions with C₆H₆-EtOAc (3:1) gave a mixture which was separated into two compounds by preparative TLC (Sigel; C₆H₆-EtOAc; 1:1). One of these was characterized as luteolin by direct comparison with an authentic sample whereas the second compound (1) was a new flavone, here named abrectorin. Et,O- and EtOAc-insoluble fractions containing glycosidic components were combined, concentrated and the resulting solvent-free concentrate was chromatographed on a Sigel column, which was eluted with EtOAc-MeOH (3:1) to give a mixture of three compounds which on preparative-PC using n-BuOH-HOAc-H₂O (4:1:5; upper layer) yielded orientin, isoorientin and a flavone glycoside (2).

Abrectorin (1)

Colour reactions and spectral data indicated 1 to be a polyhydroxyflavone. Methylation of 1 yielded a dimethyl

ether identical with 6,7,3',4'-tetramethoxyflavone (1a) [Bhardwaj, D. K. et al., unpublished results] showing that 1 was a 6,7,3',4'-dihydroxydimethoxyflavone. It gave a negative Quastel test [7] showing the absence of an orthodihydroxyl in the molecule. On alkali fission 1 yielded isovanillic acid, fixing an OH and an OMe at C-3' and C-4' respectively. The solubility of 1 in aqueous Na₂CO₃ (10"_o) indicated another OH at C-7, so that the other OMe was therefore at C-6. Abrectorin is thus 6,4'-dimethoxy-7,3'-dihydroxyflavone (1). This was confirmed by the identity of its diethyl ether with synthetic 6,4'-dimethoxy-7,3'-diethoxyflavone (1b) [8].

Desmethoxycentaureidin-7-O-rutinoside (2)

Colour reactions and spectral data indicated 2 to be a polyhydroxyflavone glycoside. On hydrolysis it yielded an aglycone (2a) and two free sugars identified as rhamnose and glucose by PC (n-BuOH-Py-H₂O, 6:4:3). Methylation of 2a yielded a trimethyl ether identical with 5,6,7,3',4'-pentamethoxyflavone (2b) [9] indicating that 2a was a 5,6,7,3',4'-trihydroxydimethoxyflavone. On alkali fission 2a yielded isovanillic acid showing a OH and a OMe at C-3' and C-4' respectively. 2a did not give the Bargellini test) for a 5.6.7-trihydroxy system or the Quastel test [7] characteristic for an ortho-dihydroxy system. Consequently, the remaining OMe and OH were at C-6 and C-7 respectively. Bathochromic shifts in UV spectrum of 2a with AlCl₃ confirmed the chelated OH at C-5 and its solubility in aqueous Na₂CO₃ (10%) supported another OH at C-7. The absence of the usual bathochromic shifts in